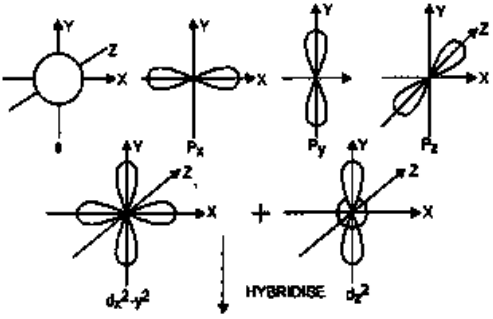
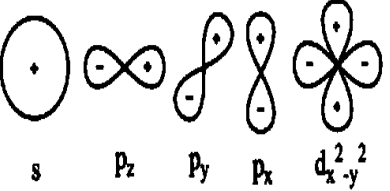
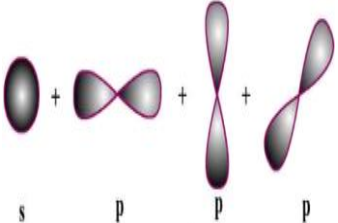
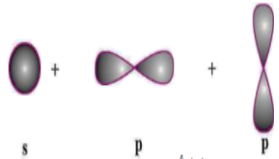
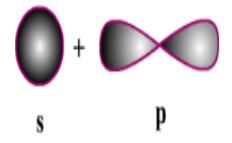
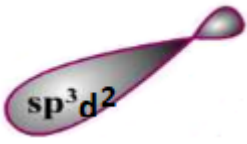
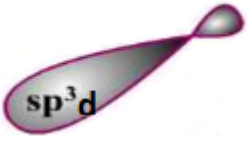



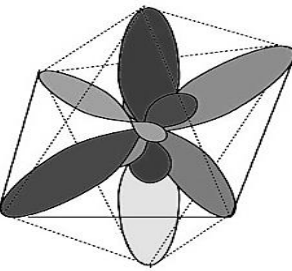
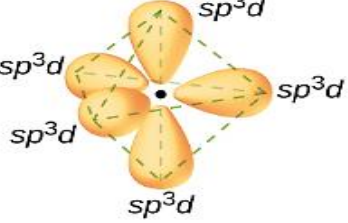
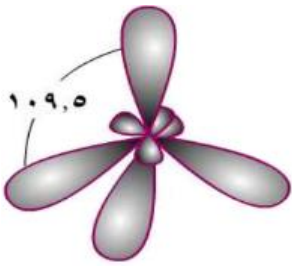
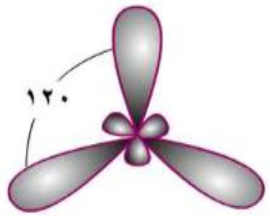
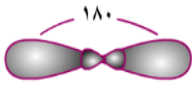


مقارنة بين انواع التهجين

Sp^3d^2	Sp^3d	Sp^3	Sp^2	SP	وجه المقارنة
 <p>Diagram illustrating the hybridization of one s orbital, three p orbitals (P_x, P_y, P_z), and two d orbitals ($d_{x^2-y^2}$ and d_{z^2}) to form six sp^3d^2 hybrid orbitals. The process is labeled "HYBRIDISE".</p>	 <p>Diagram illustrating the hybridization of one s orbital, three p orbitals (P_z, P_y, P_x), and one d orbital ($d_{x^2-y^2}$) to form five sp^3d hybrid orbitals.</p>	 <p>Diagram illustrating the hybridization of one s orbital and three p orbitals to form four sp^3 hybrid orbitals.</p>	 <p>Diagram illustrating the hybridization of one s orbital and two p orbitals to form three sp^2 hybrid orbitals.</p>	 <p>Diagram illustrating the hybridization of one s orbital and one p orbital to form two sp hybrid orbitals.</p>	<p>عدد الاوربيبتالات النقية الداخلة في التهجين</p>
 <p>6 افلاك هجينة Sp^3d^2</p>	 <p>5 افلاك مهجنة Sp^3d</p>	 <p>أربع افلاك Sp^3</p>	 <p>ثلاث افلاك Sp</p>	 <p>فلكين Sp</p> <p>عدد الاوربيبتالات المهجنة الناتجة</p>	
 <p>ثمانى وجوه</p>	 <p>ثنائى هرم ذو قاعدة مربعة</p>	 <p>رباعى الوجوه</p>	 <p>مثلث متساوي الاضلاع</p>	 <p>خطى</p> <p>الشكل الفراغى</p>	