# Environmental pollution: Sources, causes, effect and control

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# **Pollution**

Pollution may be defined as 'an undesirable change in the physical, chemical or biological characteristics of our air, water and land that may or will harmfully, affect human life, the lives of the desirable species, our industrial processes, living conditions and cultural assets, or that may or will waste or deteriorate our raw materials.

Pollution is mostly man made, but it can also be natural. Natural pollution is caused by volcanic eruptions, emission of natural gases, soil erosion, ultraviolet rays, cosmic rays etc.

# Types of Pollution

- Atmospheric (air) pollution
- Water pollution
- Soil/land pollution
- Radioactive pollution
- Noise pollution

# Air Pollution

## Definition

Presence of foreign material in the air which are harmful to man and environment.

# Sources of Air Pollution

- A. Natural
- B. Anthropogenic

**Domestic** 

Commercial

Agricultural

Industrial

Transportation related sources

# Natural sources

Natural pollutant emissions vary from one location to another, with seasonal, geological and meteorological conditions and with the type of vegetation eg. Volcanic eruptions, forest fires, dust storms etc.

#### Domestic Sources

In residential areas, domestic activities are the major causes of pollutant emissions.

Activity	Pollutants released
Space heating	CO, CO <sub>2</sub> , NO <sub>x</sub> , So <sub>x</sub> , soot, smoke (if fossil fuels are burned at the residence)
Cooking	Fats ( as solids, liquids, and vapors), particles, odors
Cleaning	Solvent vapors, dust, lint, spray can propellants
Gardening	Pesticides, fertilizers (some of which may be highly toxic)
Painting	Principally solvent vapors
Washing	Detergent particles, soap particles, lint

# Commercial Sources

- commercial sources of air pollution include the public services industries eg.,dry cleaning of clothes.
- Other include restaurants, hotels, schools, printing and painting.

# Agricultural Sources

- animal feed operations
- Cotton particles during harvesting and processing in sufficient quantities
- Pesticides and insecticides

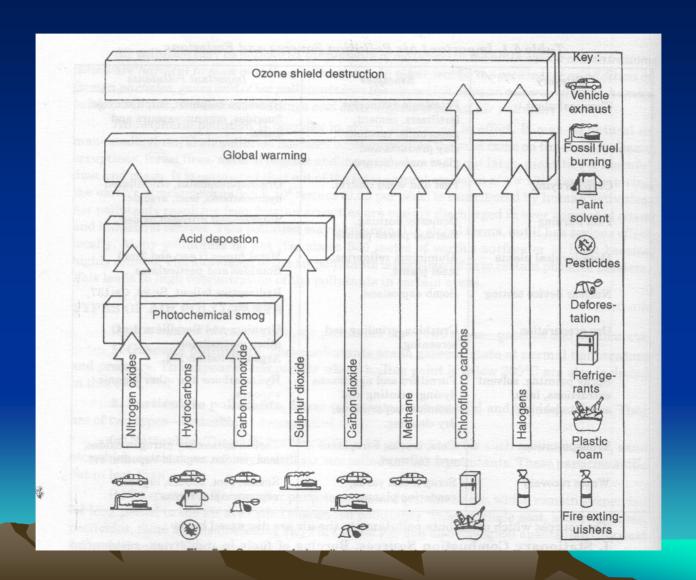
# Industrial sources

- A few of the polluting industries are
- Fertilizer and explosive manufactures (produces NO<sub>x</sub>)
- Paper plants, natural gas cleaning and processing plants, oil refineries, synthetic fibers plant (Hydrogen sulfide)
- Cast iron and other metallurgical processes (CO)
- Industries related to petroleum and natural gas industries (VOCs)

# Transportation related sources

 Except agriculture the transportation sector releases one third of the total emissions of VOCs, nitrogen oxides, and lead and more two thirds of the carbon monoxide.

# Sources of air pollution



#### Important Air Pollution Sources and emisions

Category	Examples	Important Pollutants
Chemical plant	Petroleum refineries, fertilizers, cement, paper mill, ceramic, clay products and glass manufacture	Hydrogen sulphide, sulphur oxides, flourides, organic vapors and dusts.
Crop spraying	Pest and weed control	Organophosphates, chlorinated hydrocarbons, lead, arsenic.
Fuel burning	Domestic burning, thermal power plants	Suphur and nitrogen oxides.
Metallurgical plants	Aluminium refineries, steel plants	Metal fumes ( Lead and Zinc) flourides and particulates.
Nuclear device testing	Bomb explosions	Radioactive fallouts, Sr-90, Ca-137, C-14, etc.
Ore preparation	Crushing, grinding screening	Uranium and Beryllium dust, other particulates, argon-41, lodine-131
Spray painting, solvent extractions, inks, solvent cleaning	Furniture and appliances dyeing, printings and chemical separations, dry cleaning	Hydrocarbons and other organic vapors
Transportation	Cars, trucks, aeroplanes and railways	Carbon monoxide, nitrogen oxides, lead, smoke, organic vapors etc.
Waste recovery	Scrap metal yards, rendering plants	Smoke, soot. Odors. Organic vapors metal fumes.

# **Pollutants**

Classification

Primary pollutant

Gaseous ---oxides of carbon, oxides of sulphur, oxides of nitrogen, hydrocarbons

Particulate---lead, SPM, RSPM, Pollen, dust, fly ash etc.

Secondary pollutant

Photochemical smog, acid rain

# Photochemical smog

The product of photochemical reactions are formaldehyde, peroxybenzoyl Nitrate (PBzN), peroxyacetyl nitrate (PAN), and acrolein.

The NO-NO<sub>2</sub>-O<sub>3</sub> photochemical reaction sequence

$$N_2 + O_2 \longrightarrow 2NO$$

$$2NO+O_2 \longrightarrow 2NO_2$$

In presence of sunlight

$$NO_2 + hv \longrightarrow NO +O$$

$$O + O_2 + M \longrightarrow O_3 + M$$

Where hv means photon ( $\lambda$ <0.38 $\mu$ m)

Ozone then convert NO back to NO<sub>2</sub>

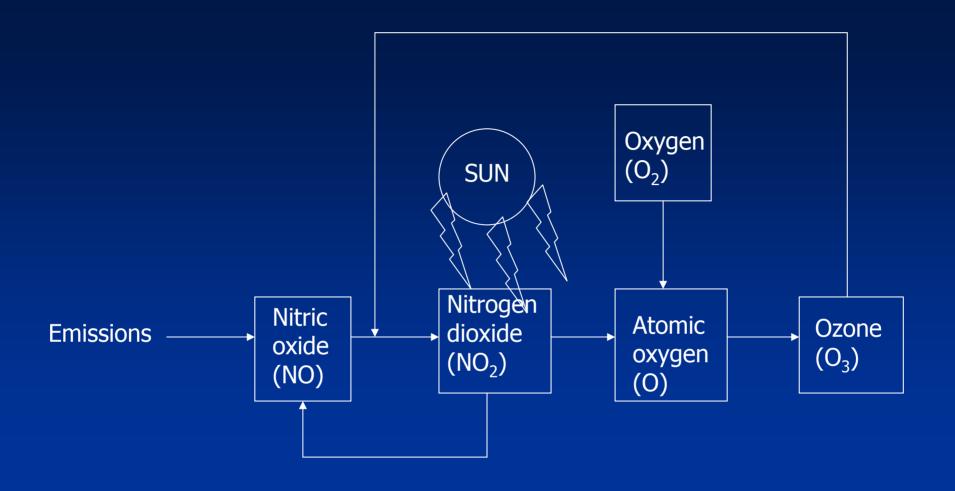
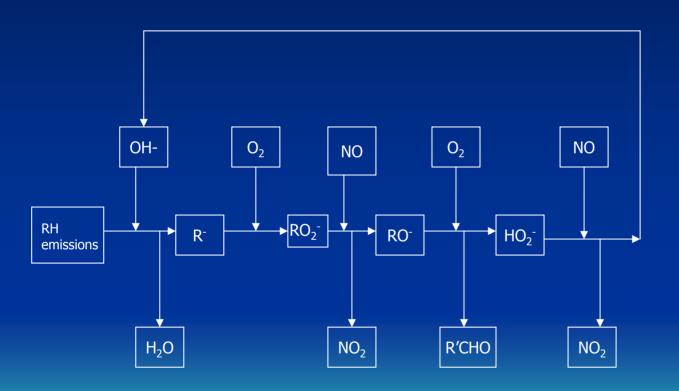


Fig. 3: Atmospheric nitrogen photolytic cycle

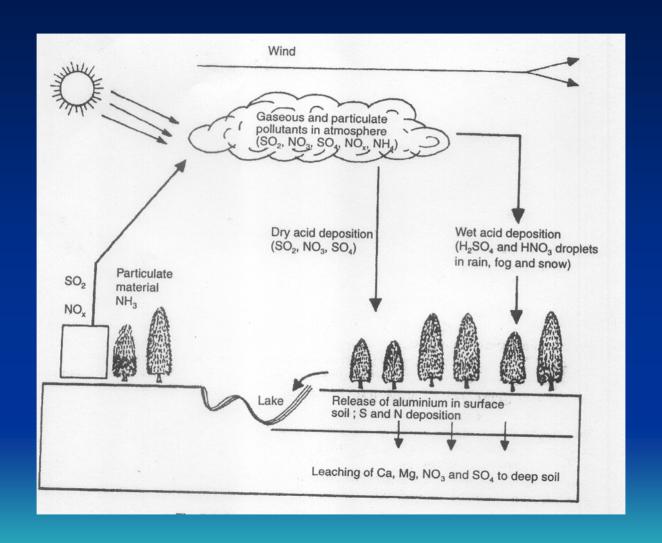
Reaction of hydrocarbons with  $NO_x$  HC can cause NO to convert to  $NO_2$  Reducing NO removes  $O_3$  slowly Increase in  $NO_2$  gives rise to increase in ozone production as shown below



#### Effects

Pollutants	Effects on Man, Vegetation and other materials
Carcinogenic hydrocarbon	On man Cancer
Carbon monoxide	On man – Poisoning, increased accident liability
Dust	On man Repiratory diseases, fiseases like silicosis (cough, cold, sneezing, allergic deseases, etc), asbestosis, byssinosis, poisoning from metallic dust
Hydrogen sulphide	On man– irritation of respiratory passages, danger of respiratory paralysis and asphyxiation On materials– Darkening of painted surfaces, corrosion
Hydrogen flouride	On man – irritation, diseases of bone (flourosis), mottling of teeth, respiratory diseases On vegetation – destruction of crops
Heavy metals	On man – specific poisoning, retardation of activities of brain, interference in enzyme activities in liver and kidney
Nitrogen dioxide	On man –irritation , brochitis, oedema of lungs
Photochemical smog (oxidants)	On man – lung irritation, asthma, bronchitis,etc On vegetation –destruction of vegetation On materials– deterioration of rubber products such as tyres and insulating wires
Sulphur dioxide	On man– suffocation, irritation of throat and eyes, respiratory diseases.
THE THE	On vegetation– destruction of sensitive crops and reduced yield On materials – corrosion.

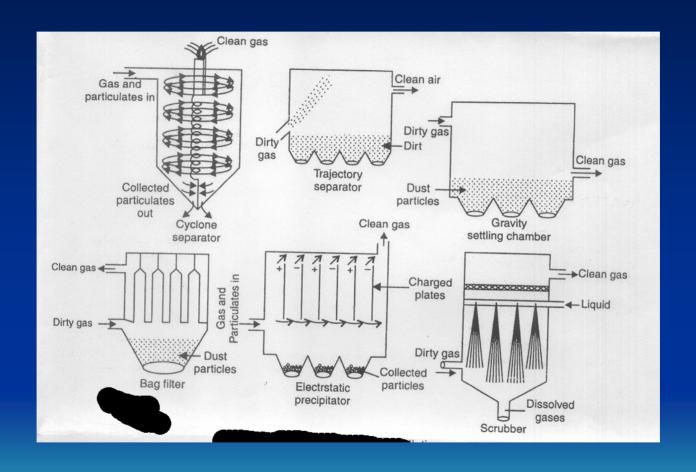
#### Acid Rain and dry acid deposition



# Control of Air pollution

- I.Control of particulate matter
- Arresters
- Scrubbers
- II. Control of gaseous pollutants
- III. Control of automobile exhaust

#### Some air pollution control devices



# Air pollution Meteorology

 The rate of decrease of temperature with increase in height is called the temperature lapse rate. If the rate of decrease refers to the air environment, it is called the environmental lapse rate.

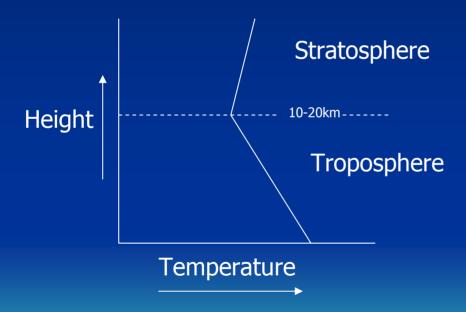


Fig.1: Temperature change with height illustrating positive lapse in the troposphere.

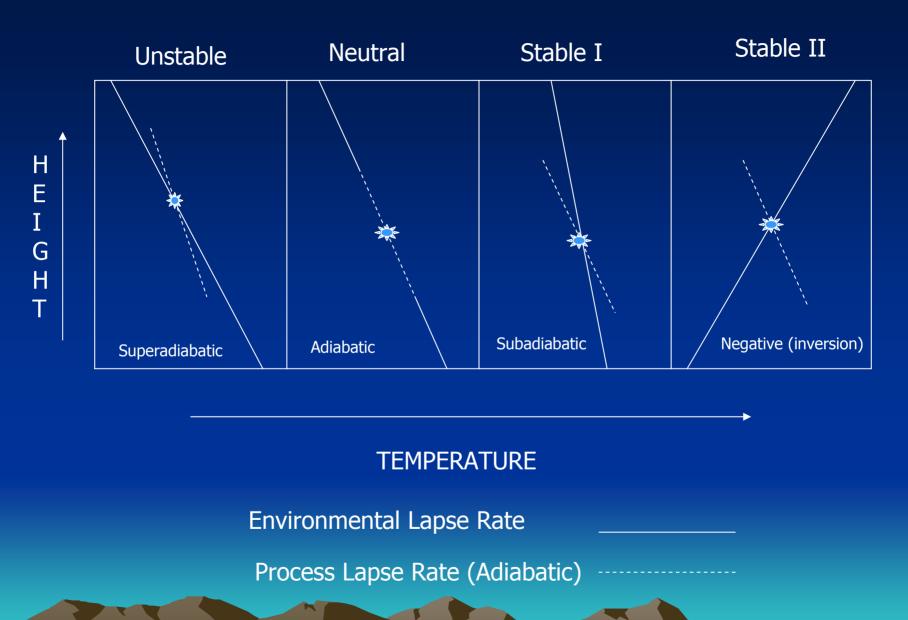
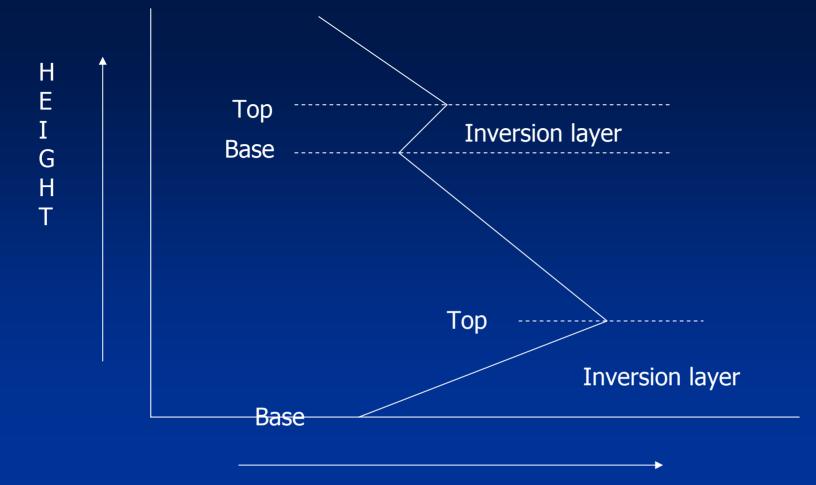


Fig 2: Stability of an air parcel determined by environmental lapse rate.



**TEMPERATURE** 

Fig.3: Surface inversion and inversion aloft.

#### Assesment

The pollution can be assesed by the use of

- a. High volume sampler
- b. Gas bubbler
- c. Respirable dust monitoring equipments
- d. Stack monitoring

# Water pollution

It is defined as the addition of some substance (organic, inorganic, biological or radiological) or factor (heat), which degrade the quality of water so that it either becomes health hazards or unfit for use.

# Sources of water pollution

#### Point source

1.Municipal waste water Most important characteristics BOD, COD,DO, metals

#### 2.Industrial waste water

Inorganic
Different metals,
Flourides, sulphates,
Cyanide, oxides of iron,
mercury,acid, alkali etc

Organic Carbohydrate, proteins, oils, fats, phenols, organic acids etc.

#### 3. Thermal pollution Main sources are thermal and nuclear power plant

Non-point sources

Agricultural sources

Fertilizers, pesticides, fungicides Substances like DDT gives rise to biomagnification

#### A few industrial sources of water pollution

Type of Industry	Inorganic pollutants	Organic pollutant
Mining	Mine Wastes: Chlorides, various metals, ferrous sulphate, sulphuric acid, hydrogen sulphide, ferric hydroxide, surface wash offs, suspended solids, chlorides and heavy metals.	
Iron and Steel	Suspended solids, iron cyanide, thiocyanate, sulphides, oxides of copper, chromium, cadmium, and mercury.	Oil, phenol and neptha.
Chemical Plants	Various acids and alkalies, chlorides, sulphates, nitrates of metals, phosphorus, fluorine, silica and suspended particles.	Aromatic compounds solvents, organic acids, nitro compound dyes, etc.
Pharmaceutical		Proteins, carbohydrates, organisolvent intermediate products, drugs and antibiotics.
Soap and Detergent	Tertiary ammonium compounds alkalies.	Fats and fatty acids, glycerol, polyphosphates, sulphonated hydrocarbons.
Food processing		Highly putrescible organic matter and pathogens.
Paper and Pulp	Sulphides, bleaching liquors.	Cellulose fibres, bark, wood sugars organic acids.

# Effect on the aquatic system

- The oxygen demand in waste consumes Do from water and it may bring it down to a value less than 4ppm which is threshold value for the survival of fish and other aquatic life. Reduced DO also eliminate the sensitive species like plankton, molusc and fish while annelid worm and some insect larvae are tolerant to low DO content.
- The DO of the water depends upon the temperature of water
- Heavy metals and high temperature also affects the organisms
- Eutrophication ————— algal bloom

# Effects on human health

- The waste water or sewage contains pathogens which when consumed causes jaundice, cholera, typhoid, amoebiosis etc.
- Heavy metals can cause serious health problems. Murcuric compounds are converted by bacterial actions into methyl mercury which causes numbness of limbs, lips and tongue, blurring of vision and metal derangement
- Minimata desease was caused in 1952 in Japan due to consumption of mercury
- Cadmium can cause itai itai disease which is a painful disease of bones and joints, cancer of lungs and liver.
- lead causes anaemia, headache, loss of muscle power and bluish line around gums

## Control

 Sewage should be suitably treated before discharging into the water body

Conventional waste treatment

**Primary** 

Screen

**Grit Chamber** 

Settling

Secondary

Biological treatment

**ASP** 

Trickling filter

Oxydation pond

Oxydation ditch

**Aerated Lagoon** 

**UASB** and

Sludge Thickening

and treatment

**Tertiary** 

Polishing ponds

Nutrient removal

Disinfection

# Soil /land Pollution

- The top soil gets polluted by the addition of the substances to the soil which adversely affect physical, chemical and biological properties of soil and reduces its productivity
- The process of soil production is very slow and hence the soil can be considered as non-renewable resource.
- Soil pollution may occur by dumping and disposing the waste directly on land, application of agrochemicals or indirectly through air pollution.
- Main soil pollutants are
- Industrial waste
- Pesticides
- Fertilizers and manures.
- Discarded waste
- Solid and semi solid
- Radioactive sources
- And other polluting materials entering indirectly

# **Effect**

- Salination of the soil
- The soil may become infertile
- The pollutants may leach down with the rain water polluting the ground water

## Control

- Manage the solid waste properly
- Pneumatic pipes should be laid for collecting and disposing waste
- Industrial waste should be dumped in special pits
- Chemical fertilizers and pesticides used may be reduced--- biofertilizer, biopesticides

# Radioactive pollution

- Natural sources--- from earth's crust
- Man made sources
- Nuclear power plant
- Atomic explosions
- Nuclear fall out
- Radioactive isotopes used in laboratories, radiation therapy

# Effects of radioactive pollution

- It depends upon half life, energy releasing capacity, rate of diffusion, rate of deposition of pollutant and the environmental factors like wind, temperature and rainfall
- The radiations can cause demage to skin, sun burns, impair metabollism etc.

#### control

- Proper precaution should be taken to prevent the nuclear fall out
- Leakage should be checked
- Safe Disposal of radioactive waste
- Regular monitoring
- Prevent occupational exposure

# Noise pollution